



We Put Science To Work

Membrane Characterization for a Sulfur-Dioxide Depolarized Electrolyzer (SDE)

Mark Elvington, Hector Colon-Mercado, David Hobbs

April 20, 2009



HyS Electrolyzer Workshop and Information Exchange

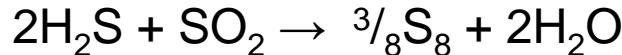
Outline



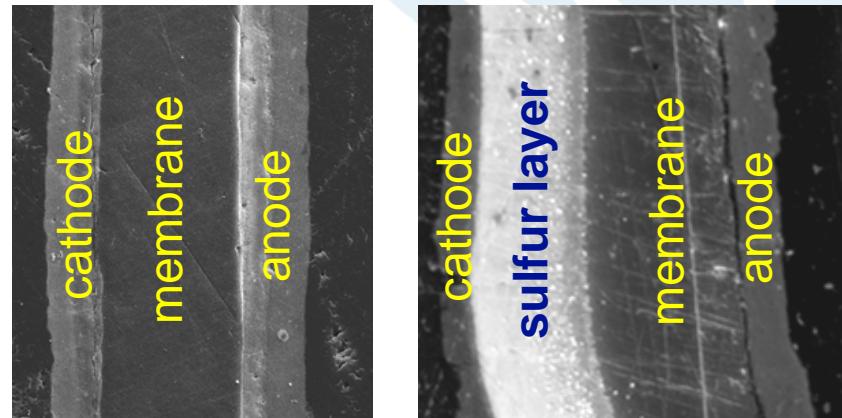
- **Background**
- **Membrane Candidates**
- **Experimental Methods**
- **Results**
 - Durability
 - SO_2 flux, SO_2 transport
 - Ionic conductivity, SDE performance

The Problem: Sulfur Layer Formation

GES has shown that the presence of both SO₂ and H₂S is required for sulfur formation (Clause reaction)



Sulfur formation is evident after multiple hours of SDE operation



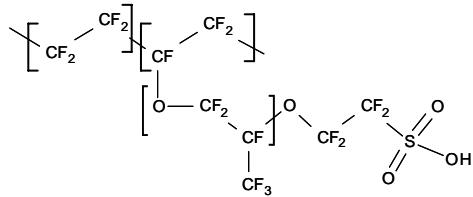
Methods for minimizing SO₂ transport

- Operating conditions: temperature, pressure, [SO₂]
- Decrease membrane SO₂ permeability
- Minimize cathode reactivity towards SO₂ through catalyst selection

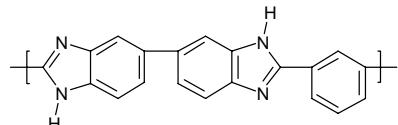
Pt: 0.77 wt%
C: 19.63 wt%
S: 75.48 wt%
F: 4.12 wt%

Membranes Tested

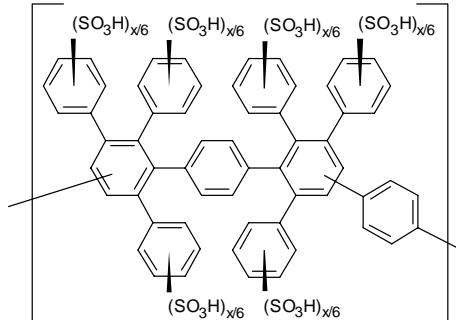
Perfluorinated sulfonic acid (PFSA)	Dupont
Polybenzimidazole (PBI)	BASF
Sulfonated Diels-Alder polyphenylenes (SDAPP)	SNL
Stretched recast PFSA	Case
PFSA/fluorinated ethylene propylene (FEP) blends	Case
Treated PFSA	GES
Perfluorocyclobutane-biphenyl vinyl ether (BPVE)	Clemson
Perfluorocyclobutane-biphenyl vinyl ether hexafluoroisopropylidene (BPVE-6F)	Clemson



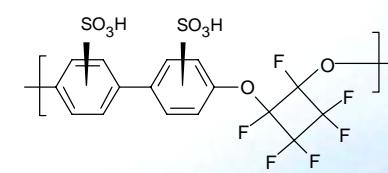
PFSA



PBI



SDAPP



BPVE-6F

Experimental Methods

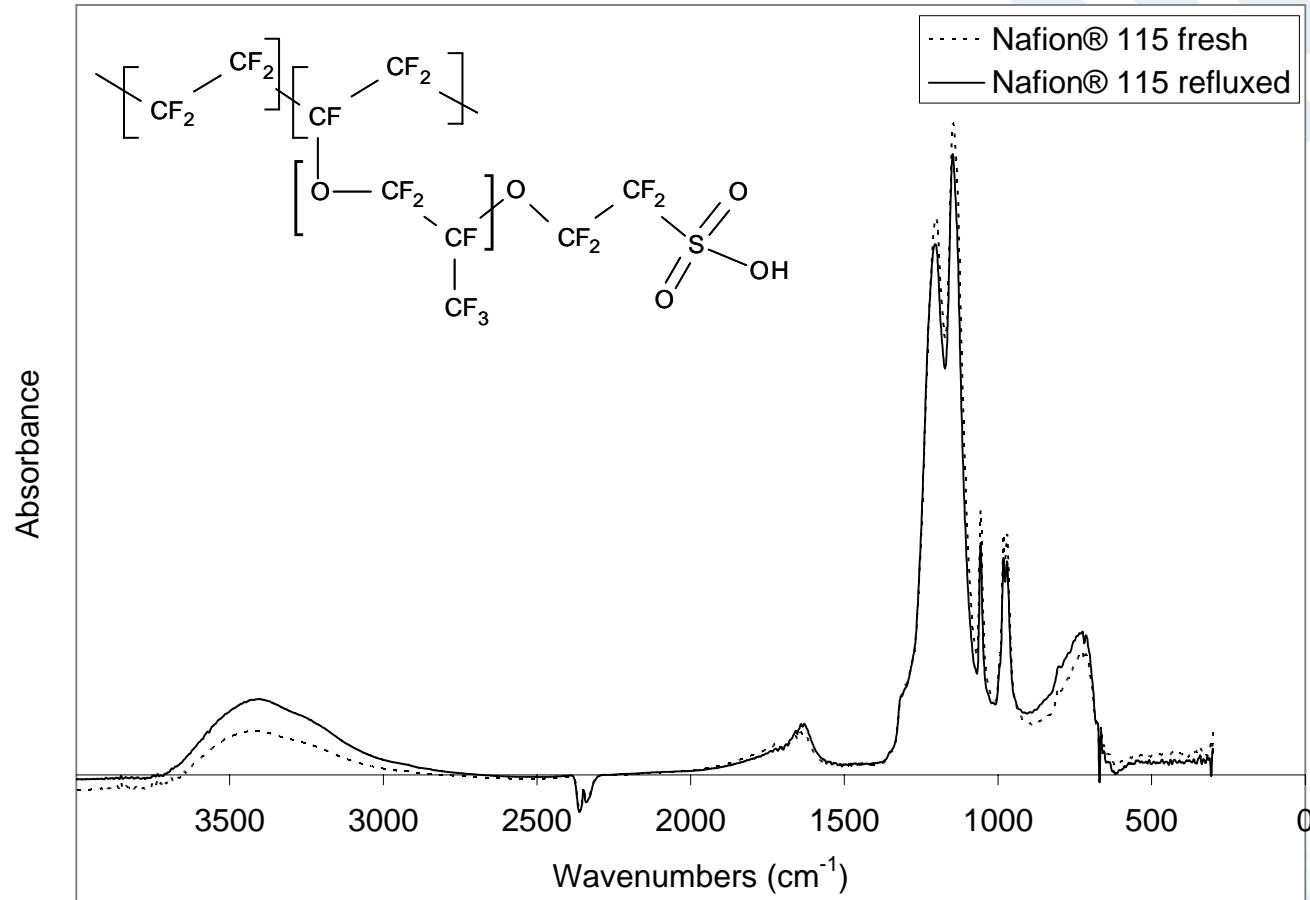
- **Chemical durability**
 - Record FTIR spectrum of membrane before and after contact with hot H₂SO₄ solution
- **SO₂ transport under non-polarized conditions**
- **Ionic conductivity**
 - Electrochemical impedance spectroscopy (EIS)
- **SDE performance**
 - Constant potential (1 V) HyS electrolyzer performance

Membrane Durability Studies

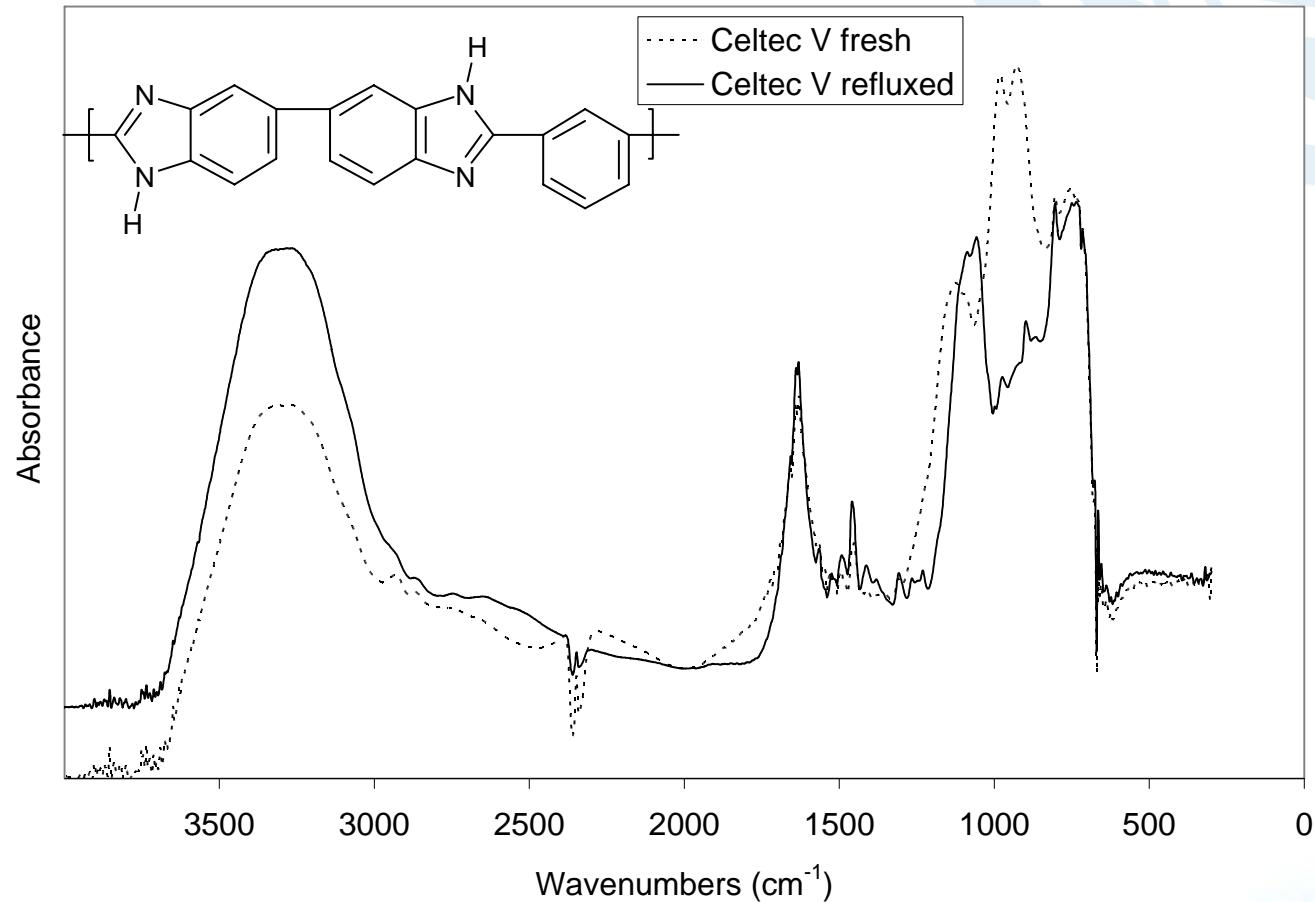
- Expose Membrane samples to 60 wt% H₂SO₄ at 80 °C for 24 hours
- Record FTIR spectra before and after acid exposure using an ATR cell
- Inspect FTIR spectra for evidence of degradation

Ir Spectra for Nafion® 115 (Dupont)

No evidence of membrane degradation



Ir Spectra for PBI Membrane (BASF)

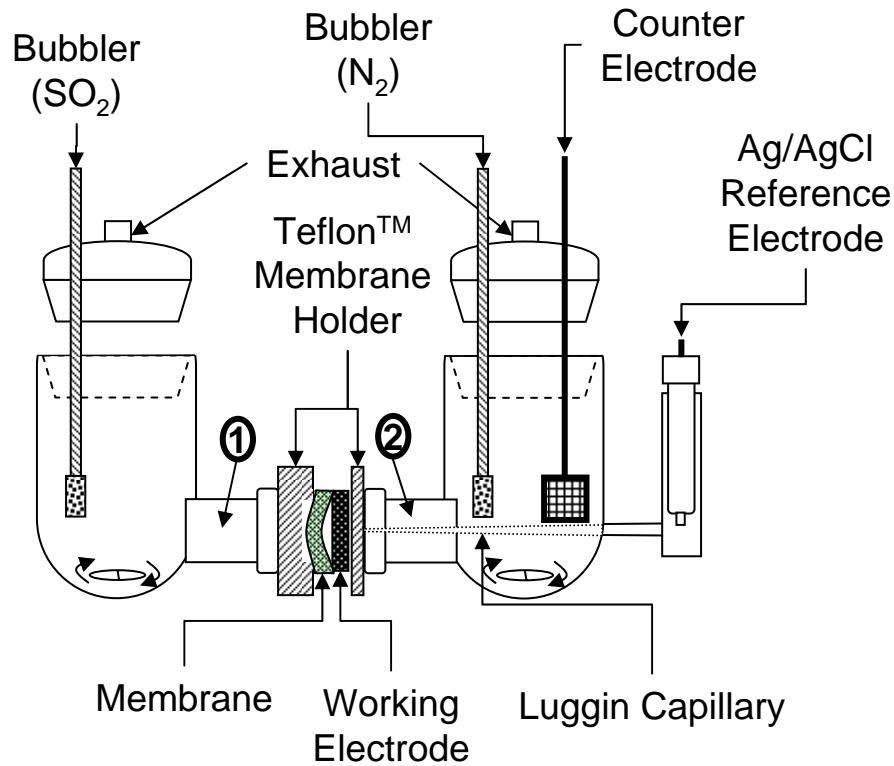


Membrane Durability Study Summary

- PFSA, SDAPP, BPVE and BPVE-6F membranes showed little or no degradation
- PBI membrane showed no degradation of the polymer backbone, however loss of H_3PO_4 was observed with simultaneous uptake of H_2SO_4

SO_2 Transport characterization Cell

Original SO_2 transport characterization cell



Original Cell Design Limitations

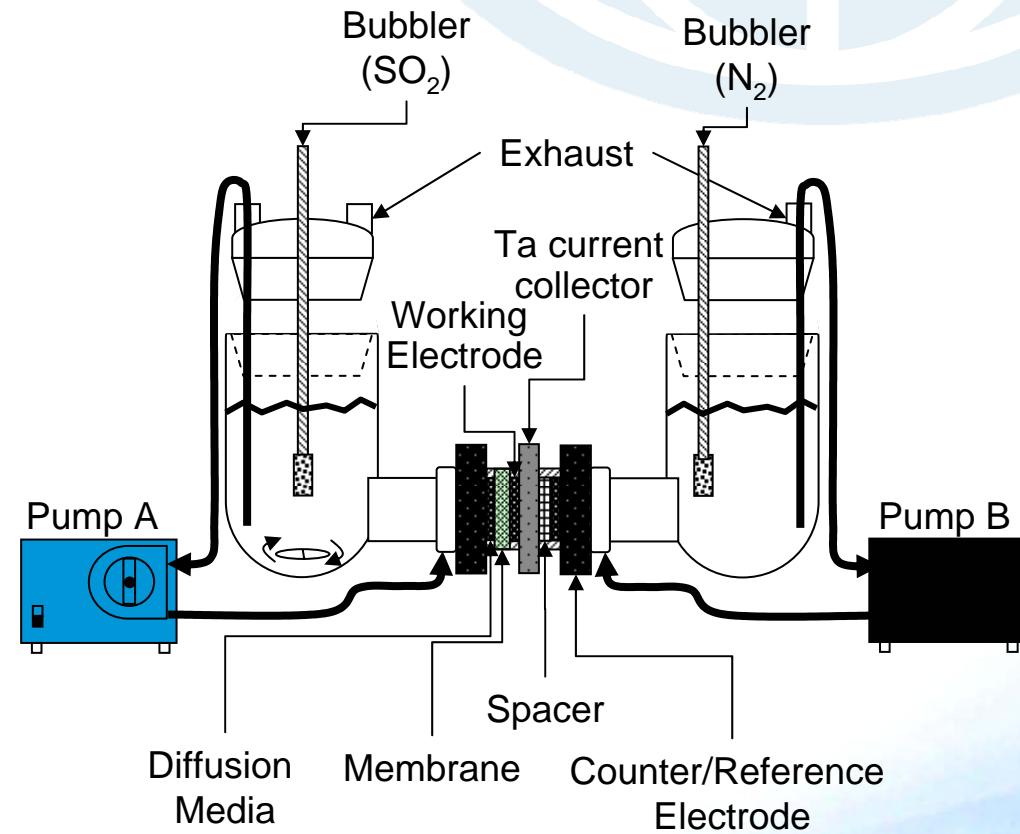
- SO_2 concentration gradient is formed in region 1
- Hygroscopic membranes can wrinkle and/or not lay flat
- H_2SO_4 generation at working electrode results in a H_2O concentration gradient in region 2

SO₂ Transport characterization Cell

Modified SO₂ transport characterization cell

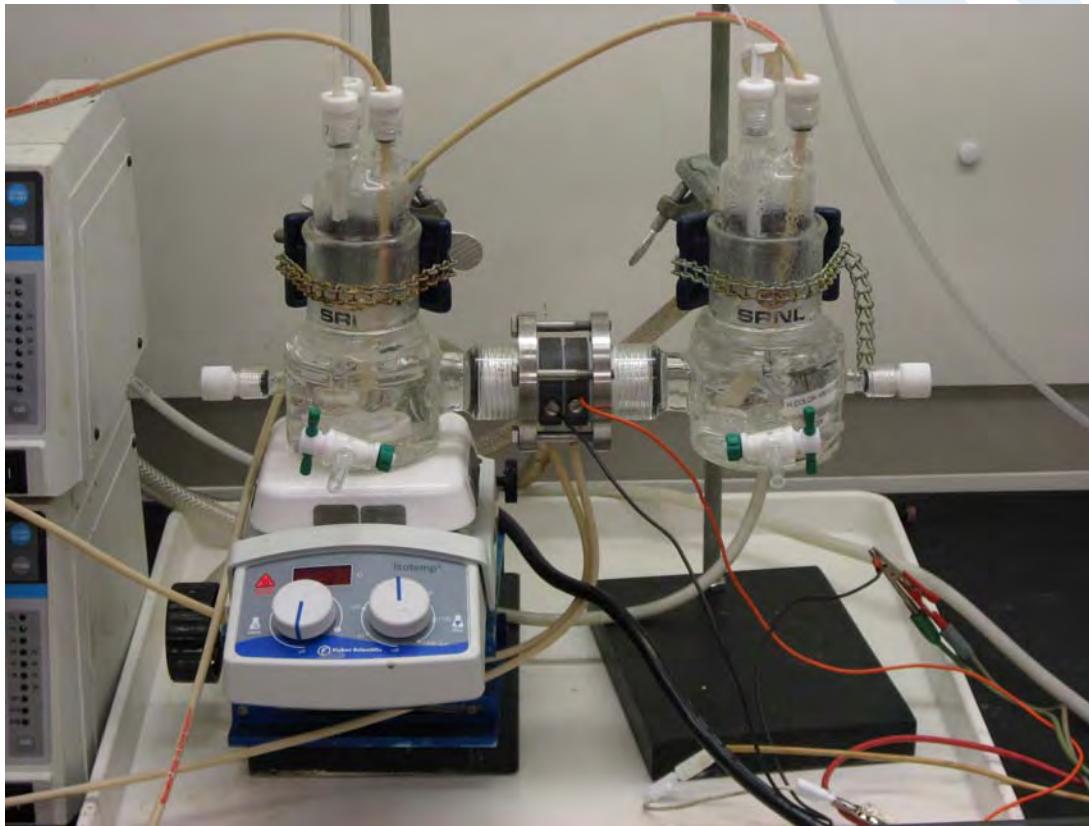
Modified Cell Design Features

- H₂SO₄ saturated with SO₂ is pumped by pump A to the membrane interface
- Graphite blocks secure and press the membrane flat
- H₂SO₄ with desired concentration is pumped by pump B into the working electrode region



SO₂ Transport characterization Cell

Modified SO₂ transport characterization cell



SO₂ Transport

SO₂ flux/transport experimental design

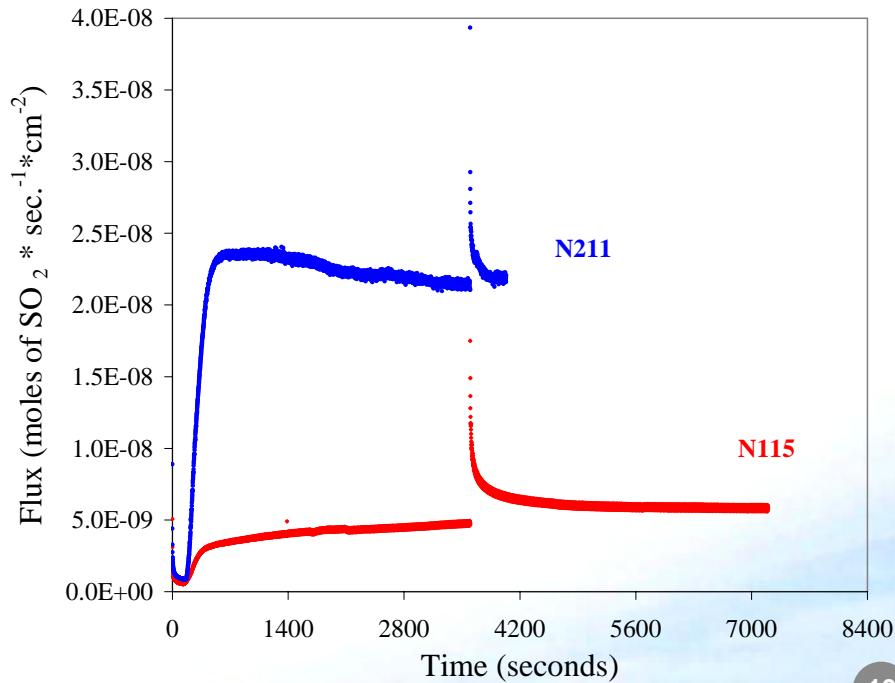
- Constant potential experiment (1 V)
- Current monitored over time in 1 hour segments
- SO₂ flux determined using Faraday's Law

$$J_{SO_2} = \frac{i}{nF}$$

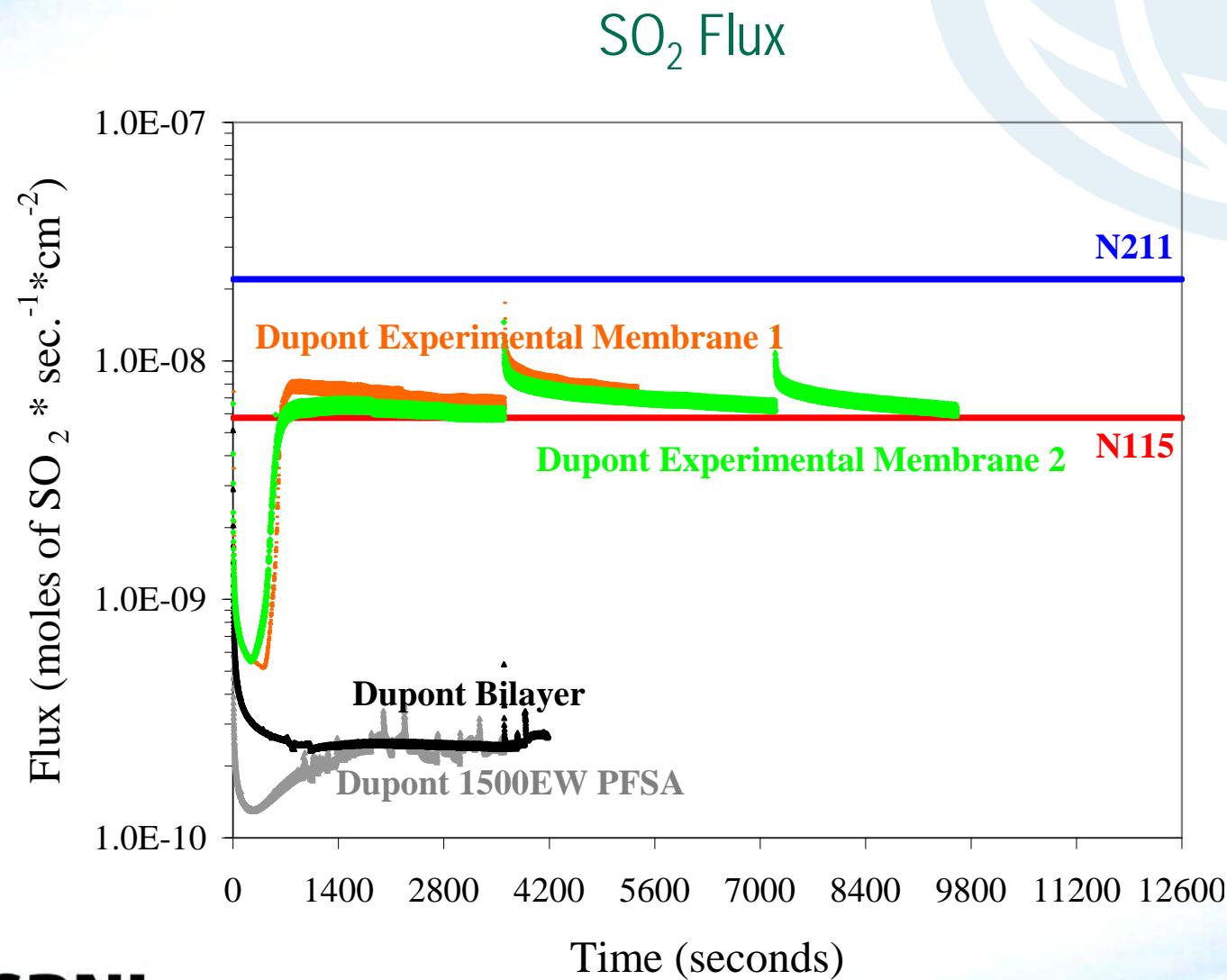
- SO₂ transport can be estimated by Fick's first law of diffusion

$$SO_2 \text{ transport} = \frac{J_{SO_2} L}{C_0}$$

- Nafion® 115 and Nafion® 211 serve as baselines



SO_2 Transport: Experimental Membranes (Dupont)



SO₂ Transport: Experimental Membranes (Dupont)

Membrane ID	Membrane Type	Thickness	i _{ave} (mA)	SO ₂ flux (mol s ⁻¹ cm ⁻²)	SO ₂ transport (cm ² s ⁻¹) x 10 ⁻⁸
Dupont Nafion® 115	PFSA	5.0	2.02	5.23 x 10 ⁻⁹	6.10
Dupont Nafion® 112	PFSA	2.0	5.25	1.36 x 10 ⁻⁸	6.34
Dupont Nafion® 211	PFSA	1.0	8.43	2.18 x 10 ⁻⁸	5.09
Dupont 1500 EW PFSA	1500 EW PFSA	4.0	0.055	1.43 x 10 ⁻¹⁰	0.13
Dupont bilayer	fluorinated sulfonic/carboxylic acid	5.5	0.042	1.09 x 10 ⁻¹⁰	0.14
Dupont experimental membrane 1	treated PFSA	2.0	2.55	6.61 x 10 ⁻⁹	3.08
Dupont experimental membrane 2	treated PFSA	3.5	2.32	6.01 x 10 ⁻⁹	4.90

SO₂ Transport: Stretched PFSA, PFSA/FEP blend (Case)

Membrane ID	Membrane Type	Thickness	i _{ave} (mA)	SO ₂ flux (mol s ⁻¹ cm ⁻²)	SO ₂ transport (cm ² s ⁻¹) x 10 ⁻⁸
Dupont Nafion® 115	PFSA	5.0	2.02	5.23 x 10 ⁻⁹	6.10
Dupont Nafion® 211	PFSA	1.0	8.43	2.18 x 10 ⁻⁸	5.09
Case sample 1	stretched PFSA	2.2	4.04	1.05 x 10 ⁻⁸	5.28
Case sample 4	stretched PFSA	2.5	7.66	1.98 x 10 ⁻⁸	11.47
Case sample 60-40-2	PFSA/FEP blend	2.4	2.27	5.88 x 10 ⁻⁹	3.35
Case sample 50-50-2	PFSA/FEP blend	2.2	2.30	5.96 x 10 ⁻⁹	3.01
Case sample 45-55-2	PFSA/FEP blend	2.1	1.58	4.09 x 10 ⁻⁹	1.99

SO_2 Transport: Treated PFSA Membrane (GES)

Membrane ID	Membrane Type	Thickness	i_{ave} (mA)	SO_2 flux ($\text{mol s}^{-1} \text{cm}^{-2}$)	SO_2 transport ($\text{cm}^2 \text{s}^{-1}$) $\times 10^{-8}$
Dupont Nafion® 115	PFSA	5.0	2.02	5.23×10^{-9}	6.10
Dupont Nafion® 211	PFSA	1.0	8.43	2.18×10^{-8}	5.09
Giner sample 1	treated PFSA	5.0	4.87	1.26×10^{-8}	14.70
Giner sample 2	treated PFSA	5.0	3.95	1.02×10^{-8}	11.92

SO_2 Transport: SDAPP Membrane (Sandia National Lab)

Membrane ID	Membrane Type	Thickness	i_{ave} (mA)	SO_2 flux (mol s ⁻¹ cm ⁻²)	SO_2 transport (cm ² s ⁻¹) x 10 ⁻⁸
Dupont Nafion® 115	PFSA	5.0	2.02	5.23×10^{-9}	6.10
Dupont Nafion® 211	PFSA	1.0	8.43	2.18×10^{-8}	5.09
Sandia 519C	SDAPP	3.0	4.30	1.11×10^{-8}	7.79

SO_2 Transport: BPVE, BPVE-6F Membranes (Clemson)

Membrane ID	Membrane Type	Thickness	i_{ave} (mA)	SO_2 flux (mol s ⁻¹ cm ⁻²)	SO_2 transport (cm ² s ⁻¹) x 10 ⁻⁸
Dupont Nafion® 115	PFSA	5.0	2.02	5.23×10^{-9}	6.10
Dupont Nafion® 211	PFSA	1.0	8.43	2.18×10^{-8}	5.09
Clemson B(2)	BPVE	0.71	8.17	2.12×10^{-8}	3.50
Clemson B1F1(1)	BPVE-6F (1:1)	0.63	6.24	1.62×10^{-8}	2.37
Clemson B2F1(3)	BPVE-6F (2:1)	0.75	6.80	1.76×10^{-8}	3.07

SO_2 Transport: PBI membrane, Celtec V® (BASF)

Membrane ID	Membrane Type	Thickness	i_{ave} (mA)	SO_2 flux ($\text{mol s}^{-1} \text{cm}^{-2}$)	SO_2 transport ($\text{cm}^2 \text{s}^{-1}$) $\times 10^{-8}$
Dupont Nafion® 115	PFSA	5.0	2.02	5.23×10^{-9}	6.10
Dupont Nafion® 211	PFSA	1.0	8.43	2.18×10^{-8}	5.09
BASF Celtec-V	PBI	4.0	0.82	2.14×10^{-9}	1.99

SO₂ Transport: Summary

Membrane ID	Membrane Type	Thickness	i _{ave} (mA)	SO ₂ flux (mol s ⁻¹ cm ⁻²)	SO ₂ transport (cm ² s ⁻¹) × 10 ⁻⁸
Dupont Nafion® 115	PFSA	5.0	2.02	5.23 × 10 ⁻⁹	6.10
Dupont Nafion® 112	PFSA	2.0	5.25	1.36 × 10 ⁻⁸	6.34
Dupont Nafion® 211	PFSA	1.0	8.43	2.18 × 10 ⁻⁸	5.09
Dupont 1500 EW PFSA	1500 EW PFSA	4.0	0.055	1.43 × 10 ⁻¹⁰	0.13
Dupont bilayer	fluorinated sulfonic/ carboxylic acid	5.5	0.042	1.09 × 10 ⁻¹⁰	0.14
Dupont exp. membrane 1	treated PFSA	2.0	2.55	6.61 × 10 ⁻⁹	3.08
Dupont exp. membrane 2	treated PFSA	3.5	2.32	6.01 × 10 ⁻⁹	4.90
Case sample 1	stretched PFSA	2.2	4.04	1.05 × 10 ⁻⁸	5.28
Case sample 4	stretched PFSA	2.5	7.66	1.98 × 10 ⁻⁸	11.47
Case sample 60-40-2	PFSA/FEP blend	2.4	2.27	5.88 × 10 ⁻⁹	3.35
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Clemson B1F1(1)	BPVE-6F (1:1)	0.63	6.24	1.62 × 10 ⁻⁸	2.37
Clemson B2F1(3)	BPVE-6F (2:1)	0.75	6.80	1.76 × 10 ⁻⁸	3.07
BASF Celtec-V	PBI	4.0	0.82	2.14 × 10 ⁻⁹	1.99

Conductivity and SDE Performance

Membrane electrode assembly (MEA) preparation

- Spray deposition method
- Ink: 75 wt% platinized carbon, 25% PFSA
- Anode loading target: $1.8 \text{ mg Pt cm}^{-2}$
- Cathode loading target: $0.8 - 0.9 \text{ mg Pt cm}^{-2}$

Conductivity

- Through-plane conductivity determined for MEAs
- Electrochemical impedance spectroscopy (EIS)

SDE performance experimental design

- Constant potential experiment (1 V)
- Current monitored over time in 1 hour segments

Conductivity and SDE Performance

Membrane ID	Membrane Type	Thickness	Conductivity (S cm ⁻¹)	Current Density (mA cm ⁻²)
Dupont Nafion® 115	PFSA	5.0	0.0241	270
Dupont Nafion® 211	PFSA	1.0	0.0159	393
Dupont 1500 EW PFSA	1500 EW PFSA	4.0	-	0.005
Dupont bilayer	fluorinated sulfonic/carboxylic acid	5.5	-	0.010
Dupont experimental membrane 1	treated PFSA	2.0	0.0036	127
Dupont experimental membrane 2	treated PFSA	3.5	0.0064	123
Case sample 50-50-2	PFSA/FEP blend	2.2	0.0034	155
Case sample 45-55-2	PFSA/FEP blend	2.1	0.0096	228
Sandia 519C	SDAPP	3.0	0.0328	286
Clemson B(2)	BPVE	0.71	0.0048	320
Clemson B1F1(1)	BPVE-6F (1:1)	0.63	0.0063	337
Clemson B2F1(3)	BPVE-6F (2:1)	0.75	0.0109	335
BASF Celtec-V	PBI	4.0	-	344

SO₂ Transport, Conductivity, and SDE Performance

Membrane ID	Membrane Type	Thickness	SO ₂ flux (mol s ⁻¹ cm ⁻²)	SO ₂ transport (cm ² s ⁻¹) x 10 ⁻⁸	Conductivity (S cm ⁻¹)	Current Density (mA cm ⁻²)
Dupont Nafion® 115	PFSA	5.0	5.23 x 10 ⁻⁹	6.10	0.0241	270
Dupont Nafion® 211	PFSA	1.0	2.18 x 10 ⁻⁸	5.09	0.0159	393
Dupont 1500 EW PFSA	1500 EW PFSA	4.0	1.43 x 10 ⁻¹⁰	0.13	-	0.005
Dupont bilayer	fluorinated sulfonic/ carboxylic acid	5.5	1.09 x 10 ⁻¹⁰	0.14	-	0.010
Dupont exp. membrane 1	treated PFSA	2.0	6.61 x 10 ⁻⁹	3.08	0.0036	127
Dupont exp. membrane 2	treated PFSA	3.5	6.01 x 10 ⁻⁹	4.90	0.0064	123
Case sample 1	stretched PFSA	2.2	1.05 x 10 ⁻⁸	5.28	-	-
Case sample 4	stretched PFSA	2.5	1.98 x 10 ⁻⁸	11.47	-	-
Case sample 60-40-2	PFSA/FEP blend	2.4	5.88 x 10 ⁻⁹	3.35	-	-
Case sample 50-50-2	PFSA/FEP blend	2.2	5.96 x 10 ⁻⁹	3.01	0.0034	155
Case sample 45-55-2	PFSA/FEP blend	2.1	4.09 x 10 ⁻⁹	1.99	0.0096	228
Giner sample 1	treated PFSA	5.0	1.26 x 10 ⁻⁸	14.70	-	-
Giner sample 2	treated PFSA	5.0	1.02 x 10 ⁻⁸	11.92	-	-
Sandia 519C	SDAPP	3.0	1.11 x 10 ⁻⁸	7.79	0.0328	286
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Clemson B2F1(3)	BPVE-6F (2:1)	0.75	1.76 x 10 ⁻⁸	3.07	0.0109	335
BASF Celtec-V	PBI	4.0	2.14 x 10 ⁻⁹	1.99	-	344

Acknowledgements

Funding: Department of Energy's Nuclear Hydrogen Initiative

- Savannah River National Laboratory

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- BASF

- Giner Electrochemical Systems

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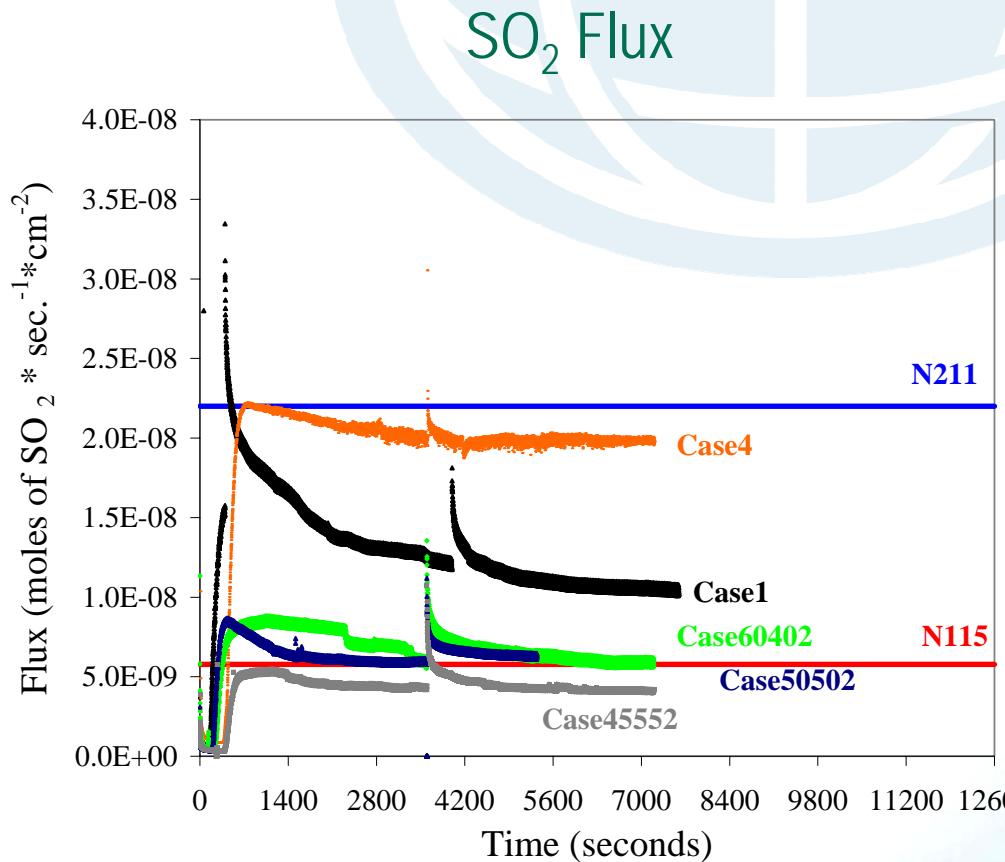
- Case Western Reserve University

Peter Pintauro



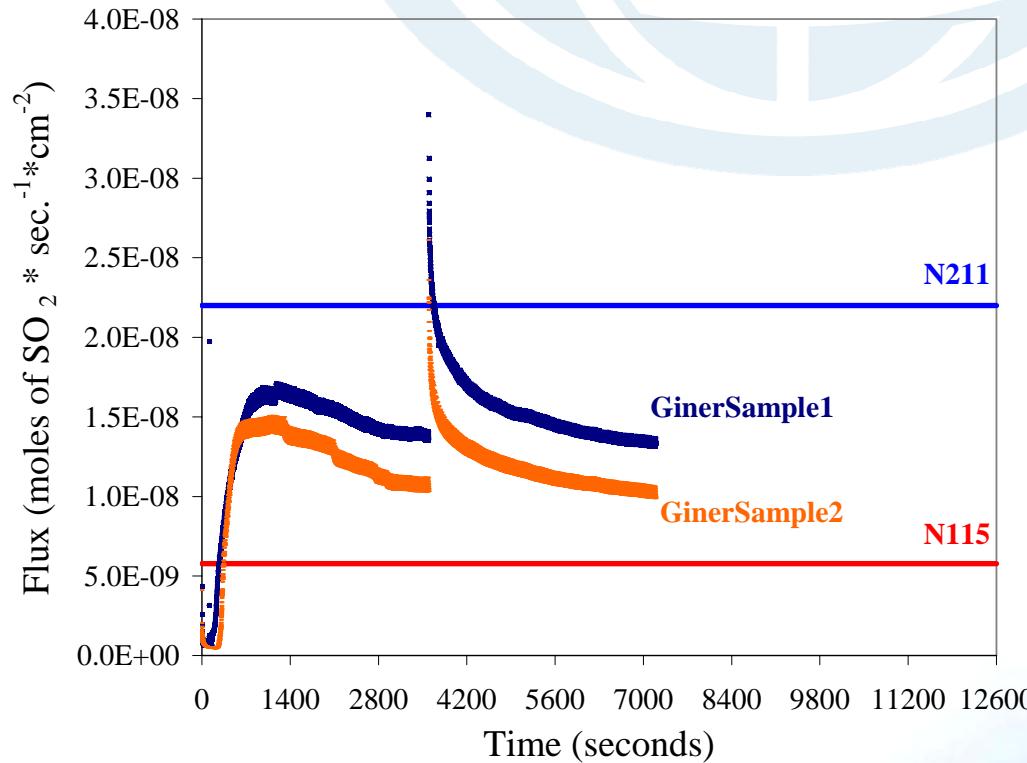
SO₂ Transport: Stretched PFSA, PFSA/FEP blend (Case)

Manufacturer and ID	Membrane type	Thickness (mil)	I _{ave} (mA)	SO ₂ flux (mol SO ₂ s ⁻¹ cm ⁻²)	SO ₂ transport(cm ² s ⁻¹) × 10 ⁻⁸
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SO₂ Transport: Treated PFSA Membrane (GES)

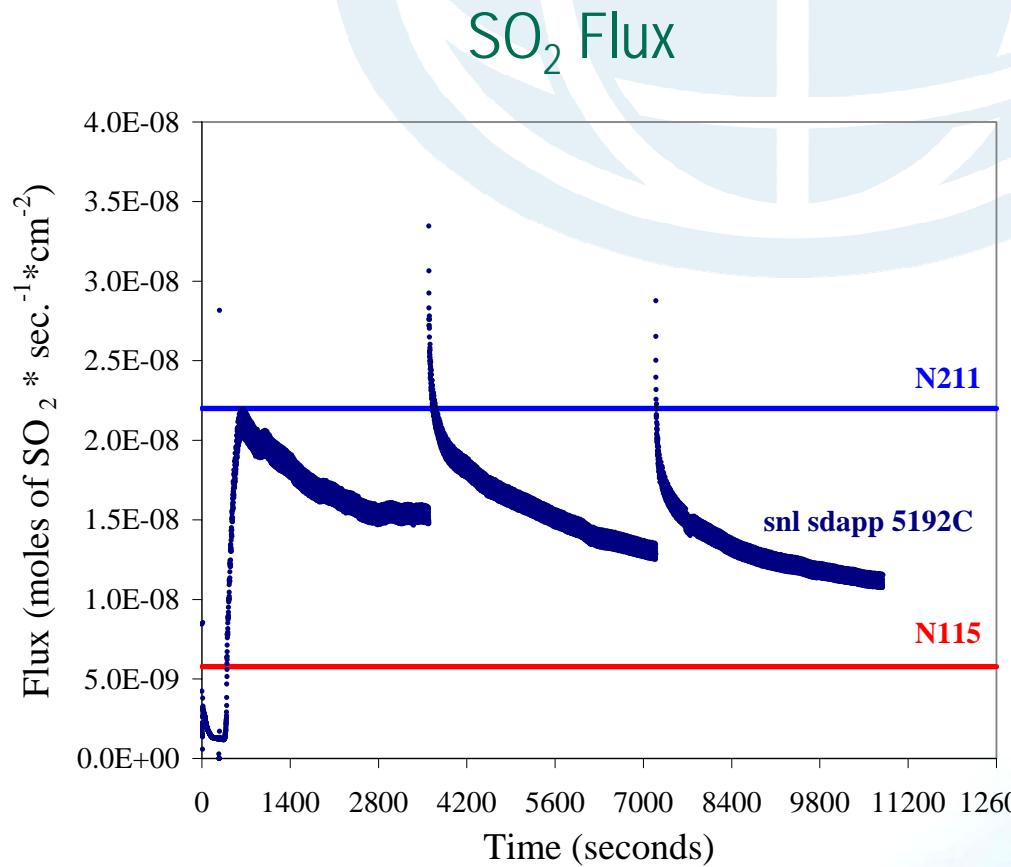
SO₂ Flux



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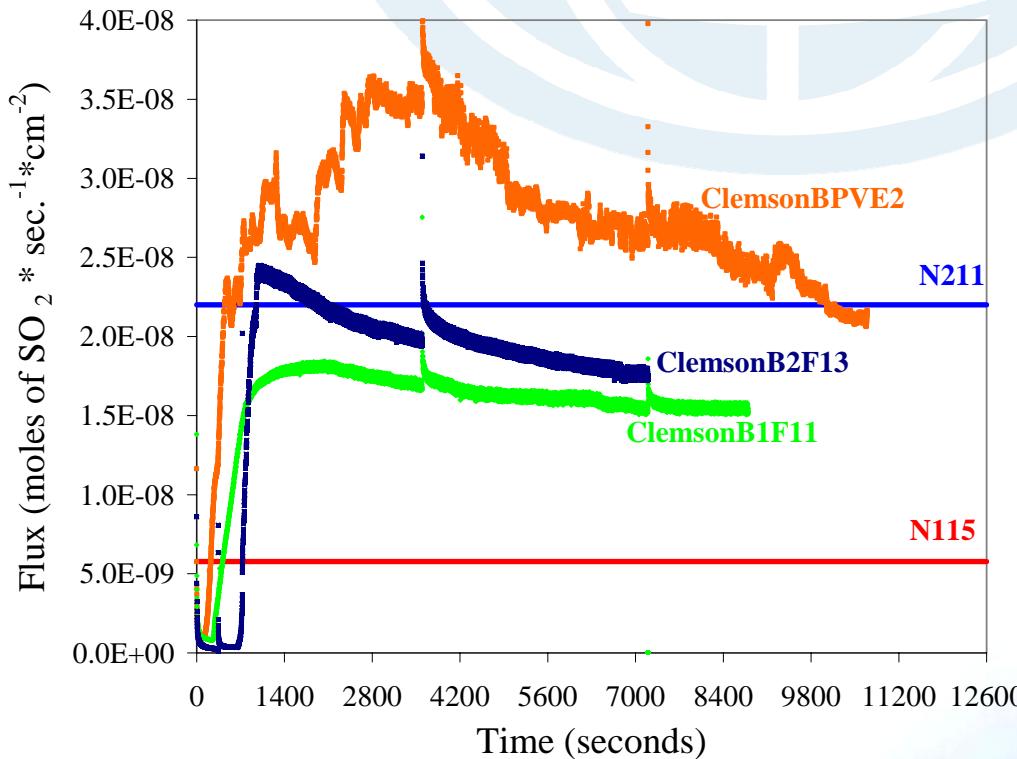
SO_2 Transport: SDAPP Membrane (Sandia National Lab)

Manufacturer and ID	Membrane type	Thickness (mil)	I_{ave} (mA)	SO_2 flux ($\text{mol SO}_2 \text{ s}^{-1} \text{ cm}^{-2}$)	SO_2 transport($\text{cm}^2 \text{ s}^{-1}$) $\times 10^{-8}$
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Dupont Nafion® 211	PFSA	1.0	8.43	2.18E-08	5.09
Sandia 519C	SDAPP	3.0	4.30	1.11E-08	7.79



SO₂ Transport: BPVE, BPVE-6F Membranes (Clemson)

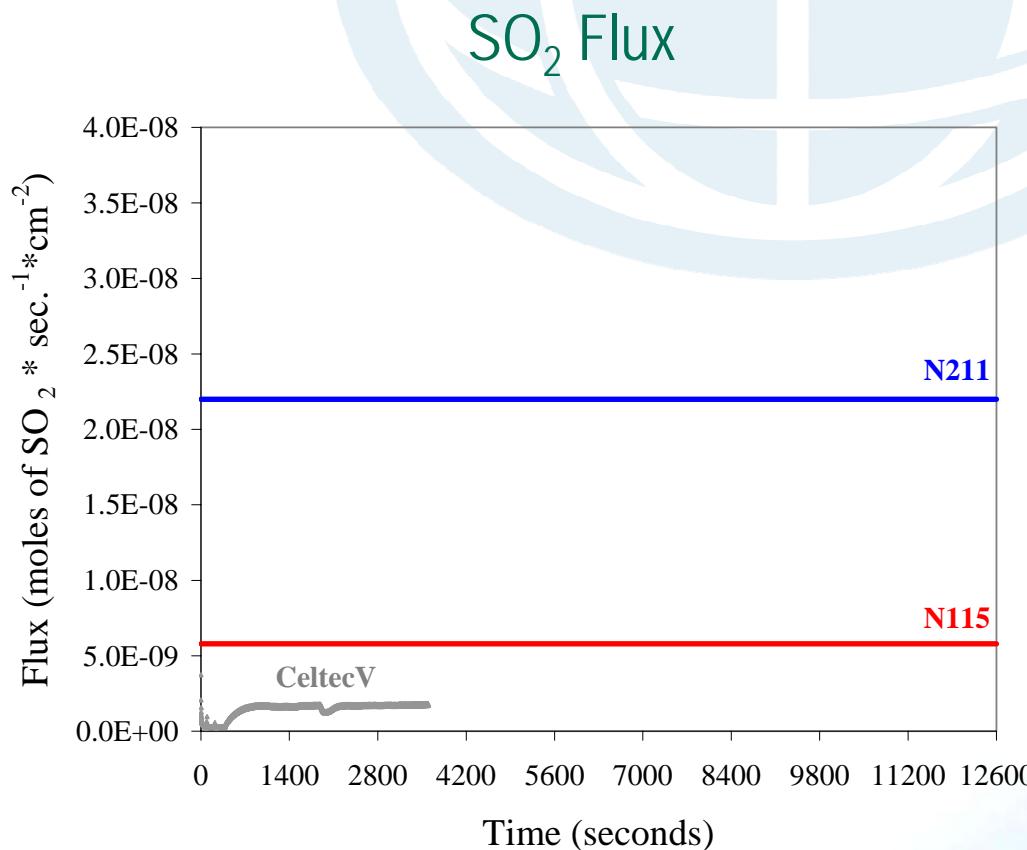
SO₂ Flux



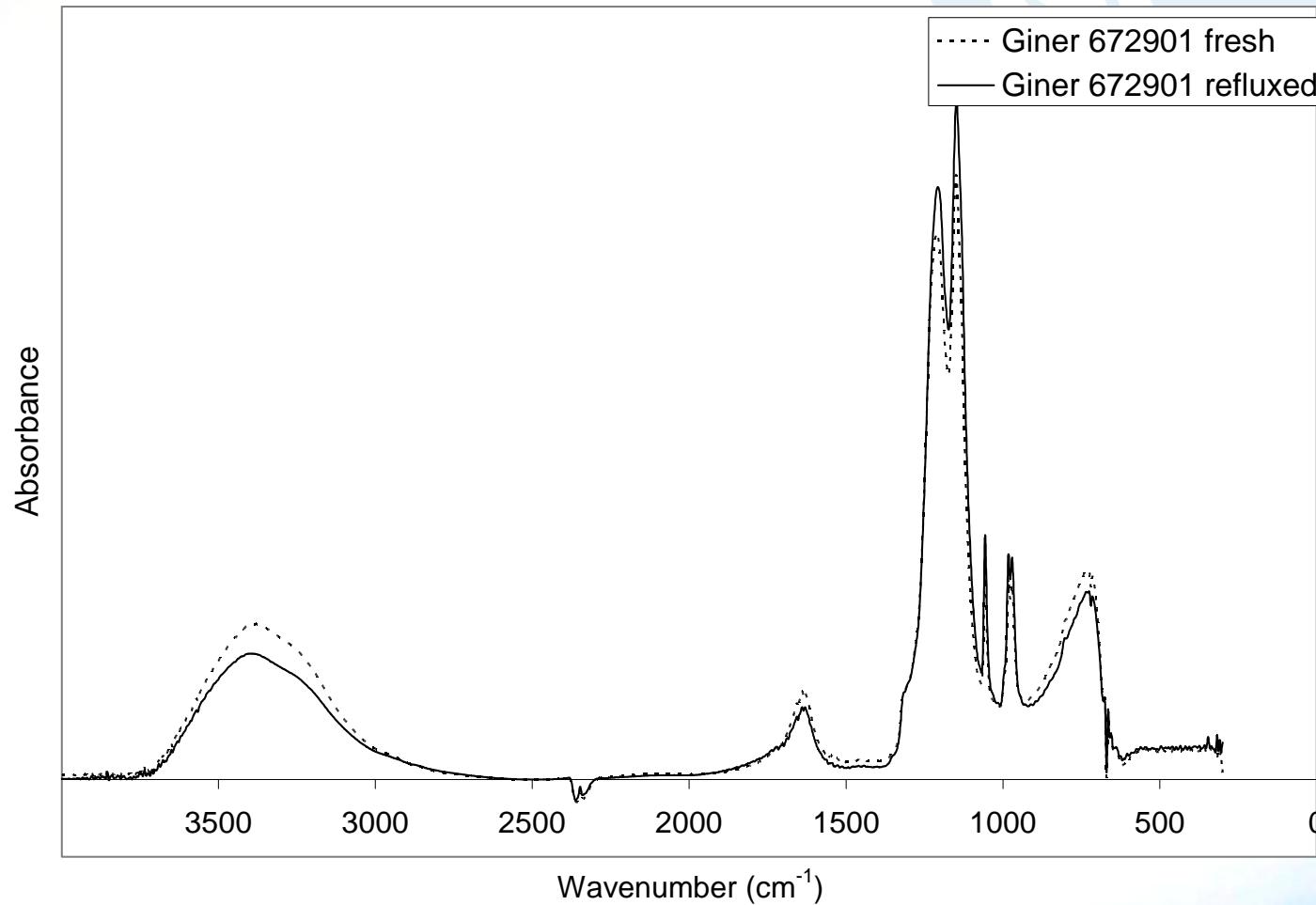
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SO_2 Transport: PBI membrane, Celtec V® (BASF)

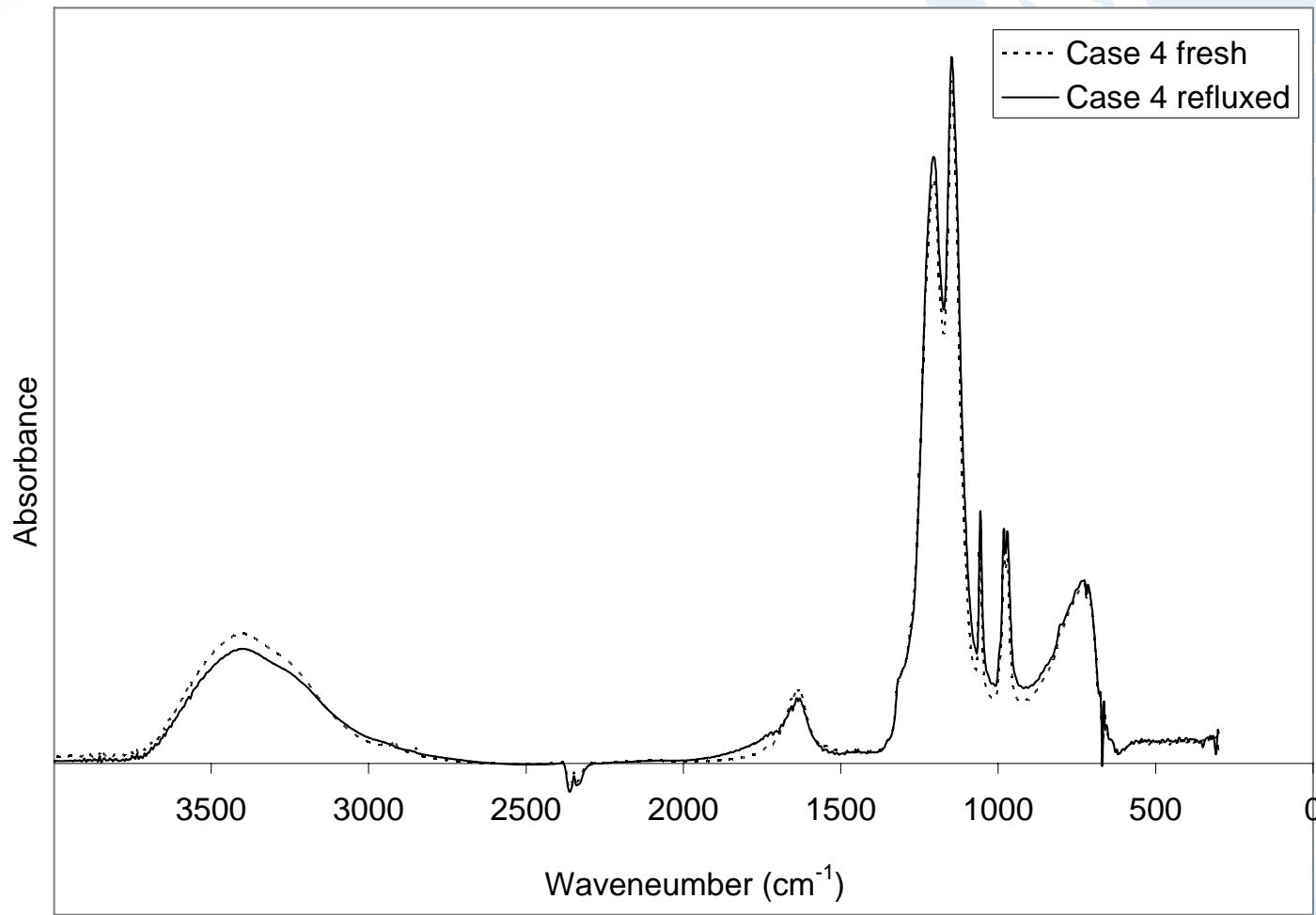
Manufacturer and ID	Membrane type	Thickness (mil)	I_{ave} (mA)	SO_2 flux ($\text{mol SO}_2 \text{ s}^{-1} \text{ cm}^{-2}$)	SO_2 transport($\text{cm}^2 \text{ s}^{-1}$) $\times 10^8$
Dupont Nafion® 115	PFSA	5.0	2.02	5.23E-09	6.10
Dupont Nafion® 211	PFSA	1.0	8.43	2.18E-08	5.09
BASF Celtec-V	PBI	4	0.824	2.14E-09	1.99



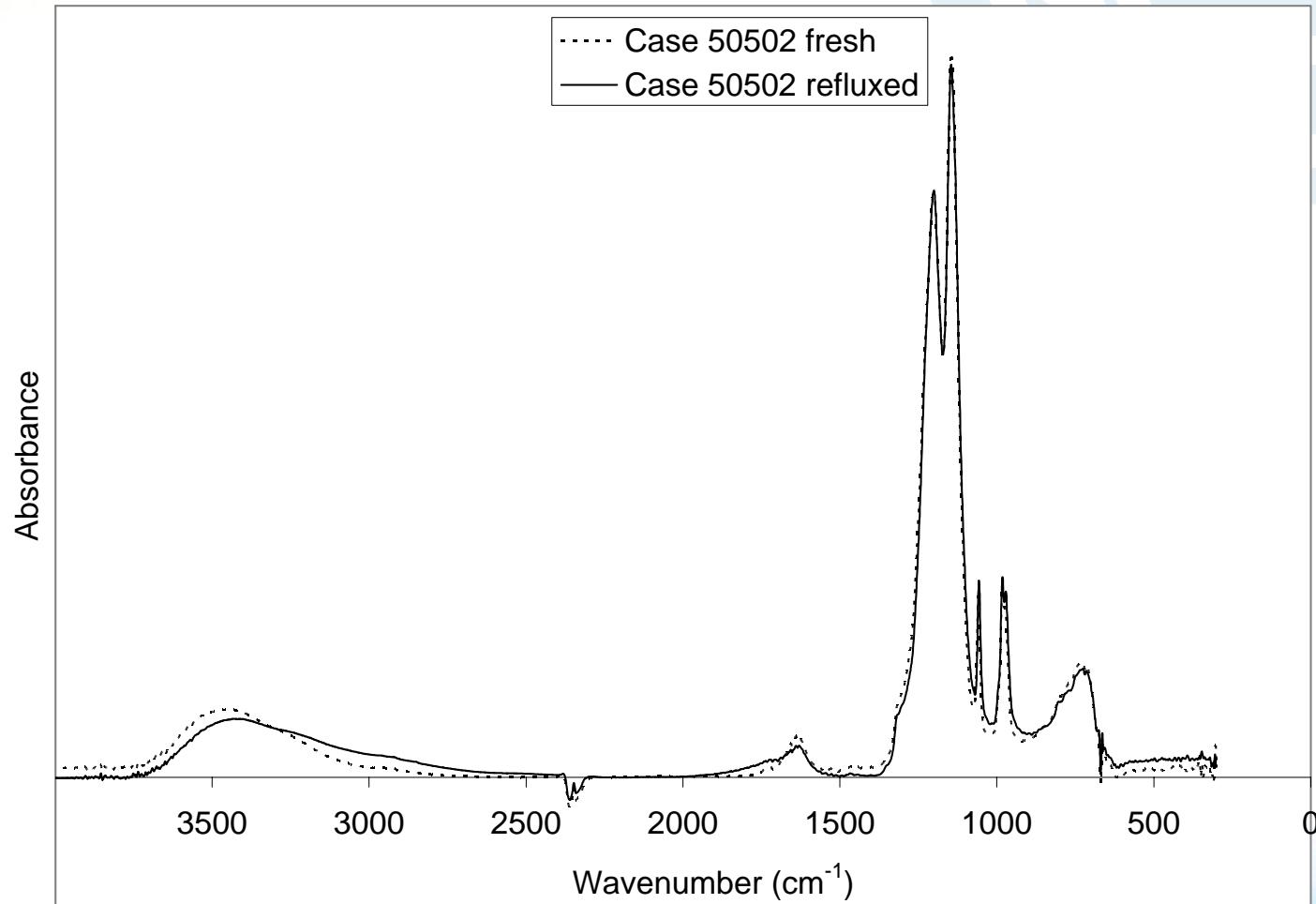
Ir Spectra for Treated PFSA Membrane (Giner)



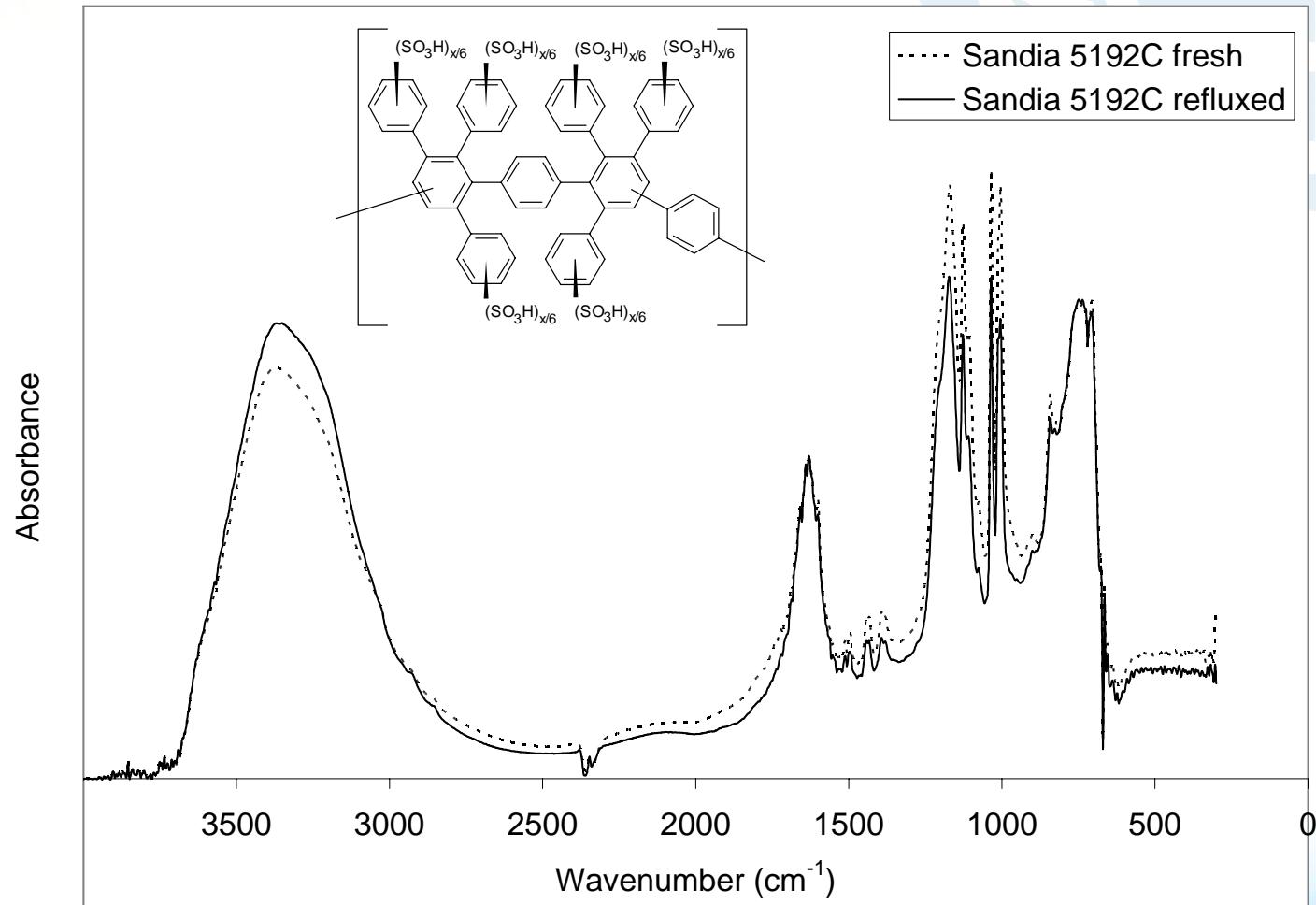
Ir Spectra for Stretched/Recast PFSA Membrane (Case)



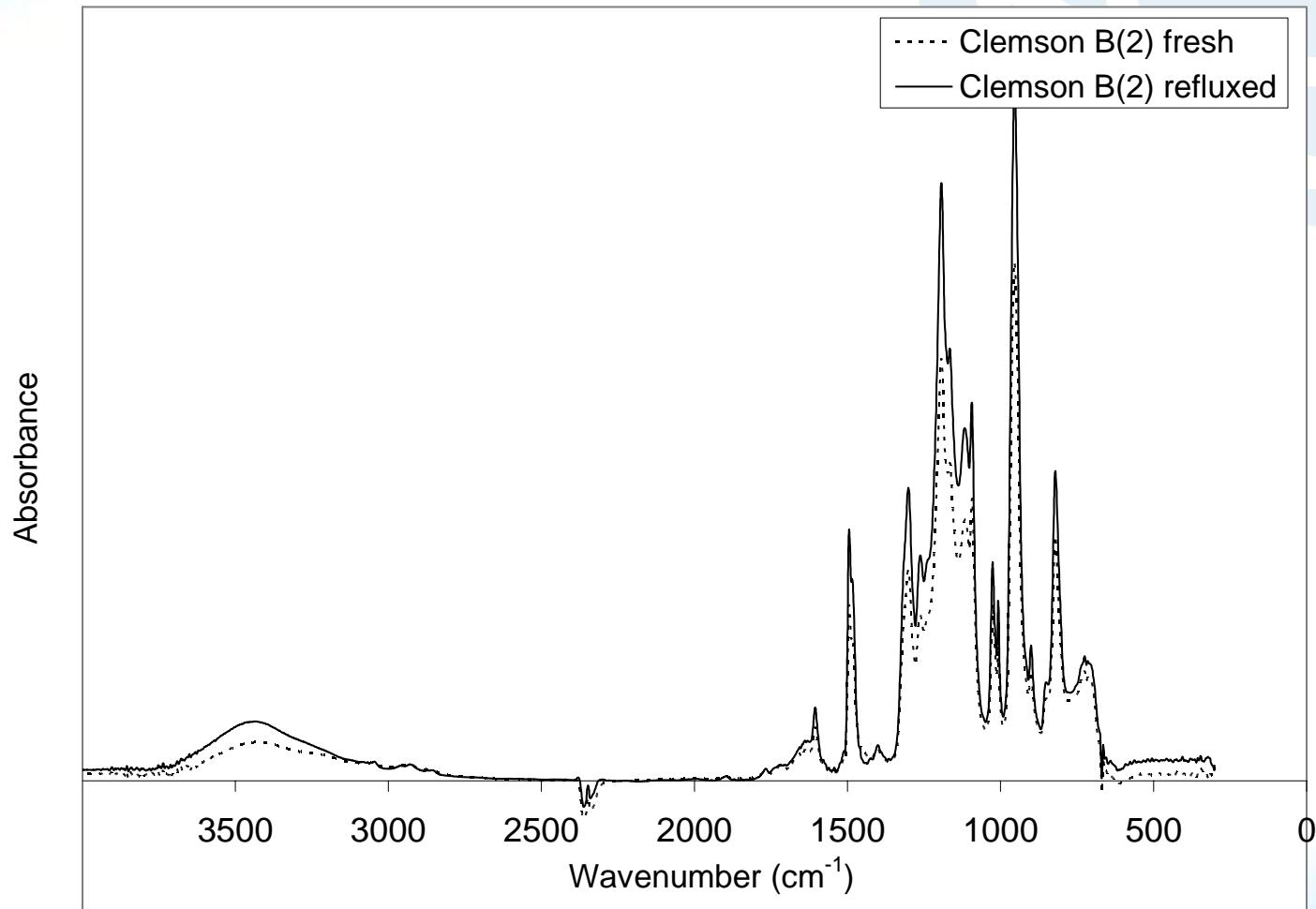
Ir Spectra for PFSA/FEP (1:1) Blend (Case)



Ir Spectra for SDAPP Membrane (SNL)



Ir Spectra for BPVE Membrane (Clemson)



Ir Spectra for BPVE-6F Membrane (Clemson)

